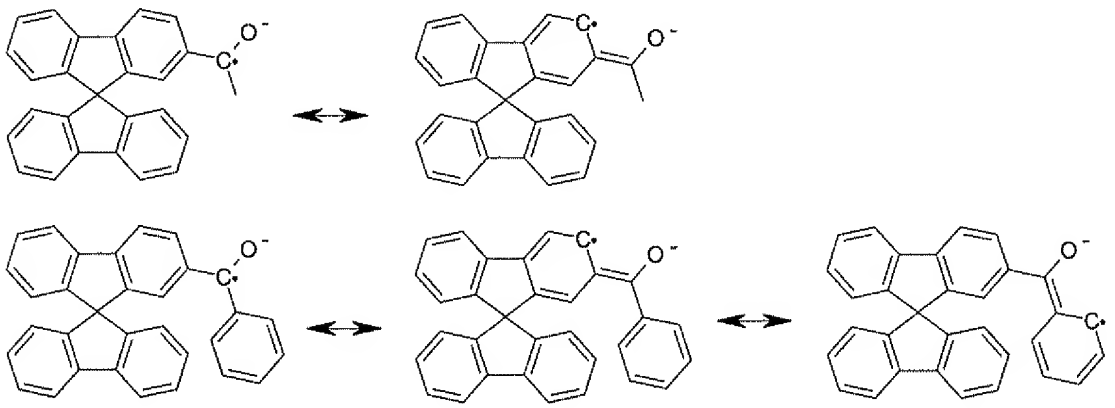
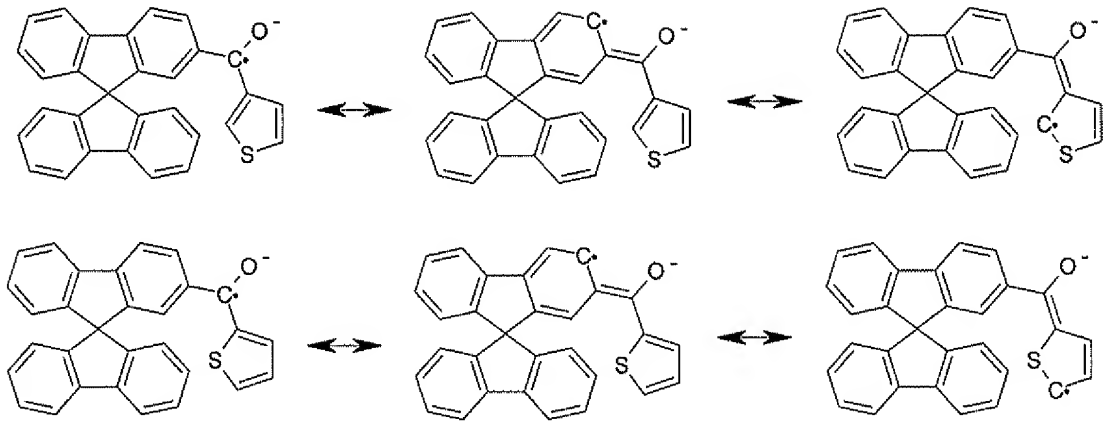
REMARKS

The applicant respectfully requests reconsideration in view of the the following remarks. In the Advisory Action the Examiner has stated that the Amendment filed April 7, 2008 does not place the application in condition for allowance. The Examiner indicated that claims 13, 14, 16 and 17 are allowable over the prior art. The applicant appreciates that the Examiner has indicated that claims 13, 14, 16 and 17 are allowable over the prior art. However, for the following reasons the applicant believes that all the claims are allowable over the prior art.

The Examiner stated that the applicant's arguments were not persuasive. In particular, he does not accept the applicant's argumentation concerning the mesomeric stabilization of all inventive compounds. The examiner is correct that the structures, which the undersigned previously submitted are flawed and contain carbon atoms with 5 bonds. The undersigned inadvertently submitted the structures without the radical. The actual structures did not contain any carbon atoms with 5 bonds, but do contain radical structures (marked by the dot in each structure). For clarity, the following structures should have been presented before:



This concept of mesomeric stabilization is valid for all inventive compounds with **any** aromatic substituent, for purely aromatic hydrocarbons as well as for heteroaromatic groups. This is shown in the following schematically for a compound wherein A is 2-thienyl or 3-thienyl (not all possible mesomeric structures are shown):



As shown above, the mesomeric stabilization is equivalent for compounds wherein A is an aromatic or heteroaromatic compound. In particular the mesomeric stabilization of the radical anion is independent of the fact that thiophene as well as pyrrol or furan are electron-rich heteroaromatic groups because the negative charge is mainly located on the oxygen of the carbonyl group and the heteroaromatic group stabilizes the radical. The assumption of the examiner that free radicals of these compounds may become localized therefore is not true.

Also the argumentation of the examiner in paragraph no. 9 of the advisory, with respect to steric shielding cannot be followed. The examiner supposes that a better steric shielding of the carbonyl group would be predicted to have negative effects, such as an increase in driving voltage. This is not the case as can be seen from examples 3 and 4 of the declaration wherein the power efficiency, which is calculated from the efficiency and from the driving voltage, is higher for the inventive compound than for the comparative acetyl compound. The higher stability of a compound can be clearly seen in the higher lifetime of the organic electroluminescent device wherein the device with the inventive compound shows more than double the lifetime compared to the device with the comparative compound.

The applicant has informed the undersigned that it is known that a compound with an acetyl group, such as the acetyl substituted spirobifluorene, can show an aldol type condensation reaction after deprotonation. This reaction is not possible with the aroyl substituted spirobifluorene. It might well be the case that the inventive compounds can contain groups, which can be deprotonated, such as hydroxy groups, but these groups do not result in any side reactions.

These arguments are valid for all inventive compounds. Therefore, the experimental results given in the declaration for a single compound combined with the argumentation as given above should be sufficient to support the full scope of the claims. In view of the above, applicant believes the pending application is in condition for allowance.

Applicant believes no fee is due with this response. However, if a fee is due, please charge our Deposit Account No. 03-2775, under Order No. 14113-00105-US from which the undersigned is authorized to draw.

Dated: May 6, 2008

Respectfully submitted,

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